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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/519,198	12/23/2004	Ken Yoshimura	Q85156	4526
23373 SUGHRUE MI	7590 10/28/200 ON, PLLC	EXAMINER		
2100 PENNSYLVANIA AVENUE, N.W. SUITE 800			HU, HENRY S	
WASHINGTON, DC 20037			ART UNIT	PAPER NUMBER
			1796	
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			10/28/2008	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)
	10/519,198	YOSHIMURA ET AL.
Office Action Summary	Examiner	Art Unit
	HENRY S. HU	1796
The MAILING DATE of this communication a Period for Reply	ppears on the cover sheet with t	the correspondence address
A SHORTENED STATUTORY PERIOD FOR REF WHICHEVER IS LONGER, FROM THE MAILING - Extensions of time may be available under the provisions of 37 CFR after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory perion. - Failure to reply within the set or extended period for reply will, by stat Any reply received by the Office later than three months after the ma earned patent term adjustment. See 37 CFR 1.704(b).	DATE OF THIS COMMUNICATION 1.136(a). In no event, however, may a reply od will apply and will expire SIX (6) MONTHS ute, cause the application to become ABANE	TION. be timely filed from the mailing date of this communication. DONED (35 U.S.C. § 133).
Status		
Responsive to communication(s) filed on Electric 2a) This action is FINAL. Since this application is in condition for allow closed in accordance with the practice unde	nis action is non-final. vance except for formal matters	
Disposition of Claims		
4) Claim(s) 1-27 is/are pending in the application 4a) Of the above claim(s) 18-25 is/are withdrest is/are allowed. 5) Claim(s) is/are allowed. 6) Claim(s) 1-17 and 26-27 is/are rejected. 7) Claim(s) is/are objected to. 8) Claim(s) 1-27 are subject to restriction and/or and/or allowed. Application Papers 9) The specification is objected to by the Examination of the drawing(s) filed on is/are: a) and an applicant may not request that any objection to the drawing of the	rawn from consideration. or election requirement. ner. ccepted or b) □ objected to by	
Replacement drawing sheet(s) including the correct 11) The oath or declaration is objected to by the		
Priority under 35 U.S.C. § 119		
12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of: 1. Certified copies of the priority docume 2. Certified copies of the priority docume 3. Copies of the certified copies of the priority docume application from the International Bure * See the attached detailed Office action for a li	ents have been received. ents have been received in Appl riority documents have been rec eau (PCT Rule 17.2(a)).	ication No ceived in this National Stage
Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date	Paper No(s)/M	mary (PTO-413) ail Date nal Patent Application

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DETAILED ACTION

1. This Office Action is in response to <u>Election</u> filed on July 21, 2008. Applicant's Election of Group I (Claims 1-17 and 26-27) without traverse is acknowledged. Applicants' two <u>IDS'</u> (1 page each) are filed so far. This Application is a 371/PCT/JP03/07704, which carries a Japanese priority date at June 28, 2002 <u>No pre-amendment is applied</u> so far. Claims 1-27 with <u>four</u> independent claims (Claims 1, 18, 23 and 24) are now pending, while non-elected three groups including Group II Claims 18-22 and 26-27), Group III (Claim 23) and Group IV (Claims 24-25) are all withdrawn from consideration. There are <u>five</u> "X" references cited in international search report for Applicants' priority document EP 1,519,435 A1 to Yoshimura. An action follows.

Claim Rejections - 35 USC § 103

- 2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

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The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

- 1. Determining the scope and contents of the prior art.
- 2. Ascertaining the differences between the prior art and the claims at issue.
- 3. Resolving the level of ordinary skill in the pertinent art.
- 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 3. The limitation of parent **Claim 1** relates to <u>a laminated membrane</u> comprising <u>two</u> membranes including:
- (A) a <u>membrane (I)</u> which comprises aromatic polymer electrolyte containing a super strong acid group and
- (B) a <u>membrane (II)</u> which comprises one electrolyte selected from the group consisting of perfluoroalkylsulfonic acid polymer electrolytes and non-super strong acid polymer electrolytes.

See other limitations of dependent Claims 2-17 and 26-27.

4. Claims 1-17 and 26-27 are rejected under 35 U.S.C. 103(a) as being unpatentable over a combination of Yushimura et al. (US 7,285,616 B2) and Hodgdon et al. (US 4,851,100).

The making of <u>laminated membrane composition</u> is achieved by comprising <u>two</u> different membranes (I and II) in parent Claim 1. In a close comparison, Membrane I is

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related to an aromatic polymer electrolyte containing a super strong acid groups as specified in Claims 5 and 6. For instance, said super strong acid directly attaching on aromatic ring is selected from four different formula such as (2a) sulfonic acid, (2b) disulfonimide, (2c) phosphonic acid, and (2d) phosphonic acid ester. Membrane II is related to a regular perfluoroalkyl sulfonic acid-containing polymer electrolyte, which said sulfonic acid group is a pendant group on the perfluoroaliphatic backbone and is thereby NOT directly attaching on the aromatic ring. A combination of two references including Yushimura and Hodgdon has obviously taught such a subject matter as following:

- 5. Yushimura et al. have disclosed the preparation of two different sulfonic acid-containing polymers to be used as polymer electrolyte membranes (PEM). One PEM membrane is that sulfonic acid group is attached on perfluoroaliphatic backbone (it is noted that such PEM is thereby equivalent to Membrane II), while the other PEM membrane is that sulfonic acid group is attached onto the ring of an aromatic polymer (it is noted that such PEM is thereby equivalent to Membrane I). See column 1, line 13-17. Yushimura has further identified the difference between such two PEM membranes, particularly heat resistance and membrane strength.

 Therefore, Yushimura is only silent about put such two different PEM membranes
 laminated together as instant parent Claim 1.
- 6. **Hodgdon** et al. have disclosed such a subject matter in the course of making a bipolar laminated membrane. For instance, **two different polarity** PEM membranes are laminated together. One membrane is related to anion exchange membrane, while the other one is related

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to cation exchange membrane (see abstract, line 1-7; column 3, line 1 – column 6, line 34).

Attention is directed to the fact that two different polarity PEM membranes are laminated together. By doing so, a constant water supply to the bipolar interface and also better stability is found on PEM membrane may be effectively achieved (see column 2, line 50-55).

- 7. In light of <u>two</u> facts that Yushimura and Hodgdon are both dealing with making and using two different PEM membranes as well as each membrane's polymer indeed carries different polarity due to chemical structure, one having ordinary skill in the art would therefore have found it obvious to <u>combine Yushimura's two different polarity PEM</u>

 membranes so as to form a laminated membrane as taught by Hodgdon. One would expect that such a combination succeed based on many similarities are existed. Additionally, the final laminated membrane product may be more effective, particularly in its water supply to bipolar interface and its stability.
- 8. The disclosure of Yushimura is found to be the same as limitations of dependent Claims 2-17 and 26-27, see specification and particularly see its application to the area of fuel cell (column 24, line 38-49). For instance, Claims 2-4 relate to structure of polymer backbone, Claims 5-7 relate to the super strong acid group, Claims 8-11 relate to –(A-Z)_m- monomer unit from Claim 7, while Claims 16-17 relate to Membrane II. Therefore, dependent Claims 2-17 and 26-27 are rejected with a combination of Yushimura and Hodgdon along with the references cited therein.

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Conclusion

9. The prior art made of record and not relied upon is considered pertinent to applicants' disclosure. The following references relate to a laminated membrane comprising two different membranes (I and II) as specified:

US 7,128,993 B2 to Barnwell et al. only discloses the making of some composite (laminated) membranes comprising: (A) <u>at least one</u> ion-conducting polymer and (B) a network of randomly oriented individual fibers. See abstract, line 1-4; column 1, line 34-38. Although the laminated structure may be formed by several layers, the claimed membrane I by using aromatic polymer is not disclosed or suggested. See column 3, line 38 – column 4, line 47. Therefore, Barnwell fails to teach or fairly suggest the laminated membrane of present Claim 1.

10. Any inquiry concerning this communication or earlier communication from the examiner should be directed to **Dr. Henry S. Hu** whose telephone number is <u>(571) 272-1103</u>. The examiner can be reached on Monday through Friday from 9:00 AM –5:00 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Dr. Vasu Jagannathan, can be reached on (571) 272-1119. The fax number for the

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organization where this application or proceeding is assigned is (571) 273-8300 for all regular

communications.

Information regarding the status of an application may be obtained from the Patent

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Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

/Peter D. Mulcahy/

Primary Examiner, Art Unit 1796

/Henry S. Hu/

Examiner, Art Unit 1796

October 26, 2008